Isolation of the Strained Bridgehead Enol Ether of
 a Bicyclo[3.3.1]nonan-2-one System †

Kazumasa WAKAMATSU, Hiroaki TAN, Norikazu BAN, Naotaka UCHIYAMA, Haruki NIWA, and Kiyoyuki YAMADA\*

Department of Chemistry, Faculty of Science, Nagoya University, Chikusa, Nagoya 464

8-Methyl-2-(trimethylsiloxy)tricyclo[5.3.1.0<sup>3,8</sup>]undec-2-ene (11) was prepared and characterized, which is the first example of the isolable strained bridgehead enol ether of a bicyclo-[3.3.1]nonan-2-one system. The reactions of 11 with electrophiles readily occurred at the C-3 bridgehead position.

Since the preparation of bicyclo[3.3.1]non-1-ene by Marshall<sup>1a)</sup> and Wiseman<sup>1b)</sup> in 1967, intensive studies on the synthesis of strained bridgehead olefins have been made.<sup>2)</sup> Although the decarboxylation of certain bicyclic  $\beta$ -keto acids such as the acid (1) is believed to proceed through the strained bridgehead enols, 1,3) the isolation of such intermediates has never been reported. The transient existence of the bridgehead enol ether of bicyclo-[3.3.1]nonan-2-one (3) in the thermolysis of the ester (2) was suggested by Bloch.<sup>4)</sup>

In connection with the synthesis of sesquiterpenes, we have studied the reactivities of the bridgehead enolates of several bicyclo[3.3.1]nonan-2-one compounds. We describe herein the synthesis, characterization, and reactions of 8-methyl-2-(trimethylsiloxy)tricyclo[5.3.1.0<sup>3</sup>,<sup>8</sup>]undec-2-ene (11): this is the first example of the isolable bridgehead enol ether of the bicyclo[3.3.1]nonan-2-one system. Initially, we have attempted to obtain the bridgehead enol ether (4) of 3,3-dimethylbicyclo[3.3.1]nonan-2-one (5).5,6) On treatment of the ketone (5) with LDA (THF, -20 °C) followed by reaction with Me<sub>3</sub>SiCl (-8 °C  $\rightarrow$  room temp), the expected enol ether (4) was not formed, but there was obtained the  $\alpha$ -trimethylsilyl ketone (6),7) mp 62-63 °C (63%). By the stereoelectronic reasons, the bridgehead enolate is expected to be more stable in a bicyclo[3.3.1]nonan-2-one containing the cyclohexanone ring with the locked boat form than in the corresponding bicyclo[3.3.1]nonan-2-one with the conformationally flexible cyclohexanone ring.6,8) Thus, we have attempted to prepare the bridgehead enol

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ether (11) of a tricyclic ketone  $(\frac{7}{2})$ , in which the cyclohexanone ring of the bicyclo[3.3.1]-nonan-2-one skeleton is held in the boat conformation by the twocarbon bridge. 9) The ketone (7) was treated with LDA (DME, -20 °C) and subsequently with Me<sub>3</sub>SiCl (0 °C, 10 min, then room temp, 20 min). after being diluted with cold NaHCO3 solution, was extracted with pentane. Concentration of the extracts in vacuo gave an oil, which was purified by GLC to give the enol ether  $(11)^{7,11}$  as a colorless liquid (76%). The structure of 11 was established by the spectral data. The presence of the enol ether group in 11 was revealed by the two olefinic carbon signals [ $\delta$  125.4 (s) and 152.7 (s)] in the  $^{13}$ C NMR spectrum, the band due to the double bond (1620 cm $^{-1}$ ) in the IR spectrum, and the UV spectrum [220 nm ( $\varepsilon$  5500)]. Comparison of the IR (1620 cm<sup>-1</sup>) and UV [220 nm ( $\varepsilon$  5500)] spectral data of  $\underline{11}$  with those [1685 cm<sup>-1</sup> and 210 nm ( $\varepsilon$  2980)] of the unstrained silyl enol ether  $(12)^{12}$  reveals that the carbon-carbon double bond character of 11 is significantly reduced in comparison with that of 12. ether (11) is reactive towards electrophiles. Treatment of 11 with 1% AcOD-D<sub>2</sub>O (80 °C, 10 min) afforded the monodeuterated ketone  $(8)^{7}$  (88%). The reaction of 11 with benzaldehyde (CsF, DME, 30 °C, 2 h) afforded the hydroxy ketone  $(9)^{7}$ (colorless oil, 75%) as a 1.3:1 diastereomeric mixture together with  $\frac{7}{2}$  (27%). reaction of 11 with MeI (CsF, DME, room temp, 2 h) the methylated ketone (10)<sup>7)</sup> (colorless oil, 51%) was obtained together with  $\frac{7}{2}$  (44%).

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The bridgehead olefin with this tricyclic skeleton, tricyclo[5.3.1.0<sup>3,8</sup>]undec2-ene (A) was previously prepared and characterized: N. Takaishi, Y. Fujikura, 9)

Z-ene (A) was previously prepared and characterized: N. Takaishi, Y. Fujikura, Y. Inamoto, H. Ikeda, and K. Aigami, J. Chem. Soc., Chem. Commun.,  $\frac{1975}{1975}$ , 372. A column of 6 mm x 1.5 m packed with 2% XE-60 on Chromosorb W (AW, DMCS), 110 °C, He flow rate 66 ml/min, retention time 4.0 min. IR (CCl<sub>4</sub>) 1620, 1370, 1250, 1210 cm<sup>-1</sup>; UV (95% EtOH) 220 nm ( $\epsilon$  5500); <sup>1</sup>H NMR (90 MHz, C<sub>6</sub>D<sub>6</sub>)  $\delta$  0.20 (9H, s), 1.18 (3H, s); <sup>13</sup>C NMR (22.5 MHz, C<sub>6</sub>D<sub>6</sub>)  $\delta$  0.4 (q), 22.1 (q), 24.4 (t), 26.0 (t), 26.1 (t), 28.8 (t), 32.6 (t), 36.7 (t), 40.4 (s), 40.7 (d), 46.4 (d), 125.4 (s), 152.7 (s); MS m/z 250 (M<sup>+</sup>), 235, 222. H. O. House, L. J. Czuba, M. Gall, and H. O. Olmstead, J. Org. Chem.,  $\frac{34}{2324}$  (1969). (Received September 30, 1986) 11)

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